Science 7

Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Date: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

|  |  |  |
| --- | --- | --- |
| solution | solute (in) | solvent (work) |
| dissolves | dissolving | distillation |

You can put salt in water. When you put salt in water, you make a mixture.

The mixture is called a \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_. The solution is called a salt

solution.

The water does the work. The water \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ the salt.

The water is called a \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ because it does the work of

dissolving.

The salt is put in the water. The salt is called a \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.

You can take the salt out of the water. You can separate the salt from the

water. You can take the salt out by boiling the water. Boiling the water to take out

the salt is called \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.

Draw Figure 5.6 on page 130 here. Remember to label it.

Distillation (PAGE 130)